



REVIEW

Recent Breakthroughs in Leveraging MoS₂ to Transcend the Performance Bottlenecks of Perovskite Solar Cells

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ABSTRACT: The power conversion efficiency of perovskite solar cells has increased rapidly; however, interfacial-defect-induced non-radiative recombination, ion migration, and sensitivity to moisture, oxygen, and thermal stress continue to cause open-circuit voltage losses and a limited operational lifetime. Owing to its tunable energy levels, high carrier mobility, and compact hydrophobic layered structure, two-dimensional MoS₂ can serve as an electron or hole transport layer, an interfacial buffer layer, or an additive, enabling favorable energy-level alignment and optimized interfacial charge-transfer kinetics, thereby promoting carrier extraction and suppressing recombination. The sulfur sites of MoS₂ can coordinate with undercoordinated Pb²⁺ to form Pb–S bonds, thereby passivating defects and stabilizing the α -FAPbI₃ phase, while continuous MoS₂ layers can suppress ion migration through physical diffusion blocking; meanwhile, van der Waals epitaxy and heterogeneous nucleation induced by MoS₂ improve film crystallinity and relieve residual stress, synergistically enhancing both efficiency and durability. This review summarizes the functional mechanisms, preparation methods, and integration routes of MoS₂ in PSCs, with particular attention to recent advances in wafer-scale monolayer interfaces, mesoporous electron-transport layers, functionalized nanosheet additives, quantum-dot composite absorbers, and hybrid interfaces. These approaches are compared in terms of their effects on charge transport, defect passivation, crystallization, device efficiency, stability, process compatibility, and scalability. The outlook also considers data-driven optimization, integration with other two-dimensional materials, and applications in flexible, large-area, and tandem devices.

KEYWORDS: Perovskite solar cells; MoS₂; stability; power conversion efficiency; interfacial engineering

1 Introduction

Over the past decade, MoS₂-based perovskite optoelectronic devices have attracted considerable attention [1–3]. Meanwhile, the certified power conversion efficiency (PCE) of PSCs has reached 28.0% [4], benefiting from continued advances in perovskite photovoltaic materials and interfacial engineering [5–11]. However, such “champion efficiencies”, which are typically achieved on devices with very small active areas, do not adequately reflect the practical prospects of these devices and, to some extent, obscure the key bottlenecks that hinder their technological translation. At the current stage of research, more pressing challenges than simply pushing the theoretical efficiency limit are the insufficient long-term operational stability of PSCs under realistic working conditions and the pronounced performance losses commonly observed upon scale-up [12–14]. As typical soft-lattice ionic materials, polycrystalline perovskite films inevitably develop a high density of deep-level defects at grain boundaries and at interfaces with charge-transport layers during

solution processing and rapid crystallization. Early studies often attributed device degradation primarily to external environmental factors such as moisture and oxygen. Increasing evidence, however, suggests that the coupling between trap-assisted non-radiative recombination and ion migration is a more fundamental driving force for the irreversible structural transformation of the photoactive α -FAPbI₃ phase [15,16].

In response to these intrinsic degradation pathways, a range of interfacial passivation strategies has been widely adopted [17,18], delivering clear performance gains, yet such approaches often introduce new trade-offs. Strongly coordinating molecules can effectively suppress non-radiative recombination and improve the open-circuit voltage (V_{oc}) [19,20], but their insulating nature may compromise interfacial charge extraction. Ultrathin passivation layers and trap-state regulation have therefore also been explored [21,22]. Under continuous illumination or thermal stress, some of these molecules may also desorb, thereby generating new interfacial instabilities. More importantly, strategies that perform well in small-area laboratory devices do not necessarily remain effective under scalable fabrication conditions [23–26]. Accordingly, interfacial engineering in PSCs should not be judged solely by efficiency enhancement at the laboratory scale; its effectiveness under scalable processing and its compatibility with more advanced device scenarios, including flexible electronics and tandem photovoltaics, also warrant careful evaluation. Operational stability and α -FAPbI₃ phase stabilization are also important considerations [27,28].

In light of these challenges, the development of functional materials that can simultaneously improve interfacial quality and sustain efficient charge transport has become an important direction in PSC research. Among the various two-dimensional materials explored for interfacial engineering in recent years, MoS₂ has attracted sustained attention not merely because of its general popularity as a material platform, but because it offers a comparatively balanced combination of properties across several dimensions that are particularly relevant to PSCs. Compared with representative two-dimensional materials such as graphene, which possesses an intrinsic zero bandgap and a relatively chemically inert basal plane, MoS₂ combines a moderate bandgap, a tunable electronic structure, and good compatibility with solution-based processing. These features make it particularly relevant for studies of energy-level alignment, defect passivation, and interfacial charge-transport optimization [29]. Beyond its role as a charge-transport component, MoS₂ can, under appropriate conditions, also contribute to defect passivation, regulation of interfacial energy levels, and suppression of ion migration, thereby influencing both device efficiency and operational stability. These advantages, however, are not intrinsic or universal. Their practical realization depends strongly on the structural form, Phase, mode of incorporation, and the compatibility of MoS₂ with adjacent functional layers; when integration is poorly controlled, the anticipated interfacial benefits may be offset by increased series resistance, interfacial barriers, or localized instability [30].

Structurally, MoS₂ exhibits a characteristic layered “sandwich-like” configuration, in which a plane of molybdenum atoms is sandwiched between two hexagonally arranged sulfur atomic planes. Adjacent S–Mo–S units are stacked primarily through relatively weak van der Waals interactions (Fig. 1a), a feature that facilitates conformal interfacial contact with neighboring functional layers and provides a structural basis for subsequent integration into flexible devices. In addition, the electronic band structure of MoS₂ evolves markedly with layer number: bulk MoS₂ typically exhibits an indirect bandgap, whereas the monolayer limit displays a direct bandgap (Fig. 1b). The resulting enhancement in optical absorption and pronounced excitonic effects have also made MoS₂ of continuing interest in optoelectronic applications [29]. In PSC research, MoS₂ has been explored in both conventional n–i–p and inverted p–i–n architectures. (Fig. 1c), where it can serve as a charge-transport layer, an ultrathin interfacial buffer, or a modifying component incorporated into the perovskite absorber [31]. Existing studies indicate that, under suitable configurations, these modes of incorporation can improve interfacial energy-level alignment, promote carrier

extraction, and, to some extent, suppress trap-assisted recombination at surfaces and grain boundaries [32]. At the same time, the relatively hydrophobic surface and layered stacking characteristics of MoS₂ may, under certain conditions, provide a degree of resistance to moisture and oxygen ingress, thereby improving device durability [1,3]. This effect, however, is not universal; excessive layer thickness, poor film continuity, or inadequate contact with adjacent layers may likewise introduce additional series resistance or interfacial transport barriers, thereby diminishing the expected gains [33–35].

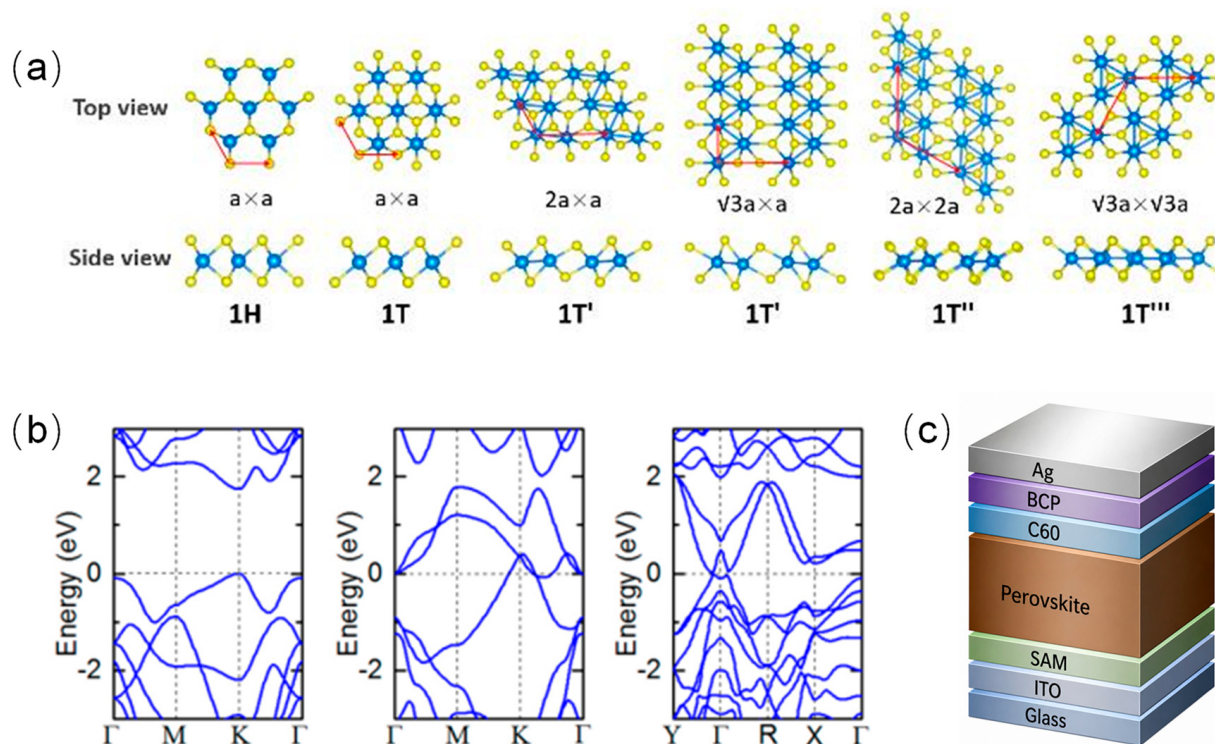


Figure 1: Structural phases, electronic band structures, and representative device architecture relevant to MoS₂ integration in PSCs. (a) Top and side views of monolayer MoS₂ with different polytypes [30], with the unit cell indicated by red arrows; (b) Band structures of 2H, 1T, and 1T' phases of MoS₂ [36]; (c) Schematic of a conventional inverted PSC architecture without MoS₂ [37].

Previous reviews have examined the structural and optoelectronic properties of MoS₂ and its applications in solar-energy devices [29], the broader use of transition-metal dichalcogenides in PSCs [31], and the interfacial engineering of halide perovskites with two-dimensional materials [38]. Recent studies published in 2024 and 2025 have extended its use beyond conventional transport and buffer layers to wafer-scale monolayer interfaces, mesoporous electron-transport layers, functionalized nanosheet additives, and quantum-dot composite absorbers. This review therefore focuses on how these approaches address the coupled efficiency, stability, and scale-up limitations of PSCs. The use of MoS₂ as an electron-transport layer, hole-transport layer, interfacial modifier, and absorber additive is compared in terms of material form, fabrication route, photovoltaic performance, stability, process compatibility, and scalability. Fig. 2 summarizes the reported PCE evolution of representative MoS₂-containing PSCs from 2016 to 2025.

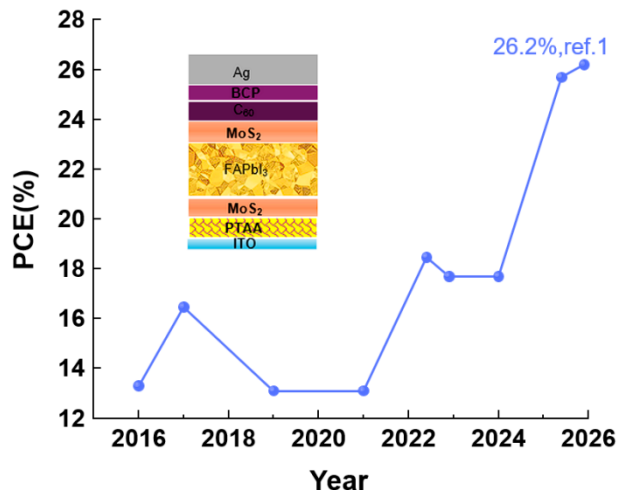


Figure 2: Evolution of the reported PCEs of representative single-junction PSCs incorporating MoS₂ from 2016 to 2025, together with schematic illustrations of representative high-efficiency device architectures.

2 Chemical Functional Mechanisms of MoS₂

The role of MoS₂ in PSCs does not arise from a single factor, but from the combined effect of a series of physicochemical interactions at the perovskite/MoS₂ heterojunction. Building on the current literature, this section discusses the main mechanisms of MoS₂ from four perspectives—interfacial charge transfer, defect passivation, crystallization control, and stability enhancement—to clarify the pathways through which it improves device performance and the conditions under which these effects are likely to hold.

2.1 Interfacial Charge-Transfer Kinetics

Efficient charge separation and rapid carrier transport are prerequisites for achieving high PCE. MoS₂ can improve interfacial charge-transfer kinetics by modulating the energy barrier at the perovskite heterojunction. Owing to its layer-dependent bandgap and tunable Fermi-level position, the electronic structure of MoS₂ can be adjusted to form a favorable cascade alignment with the perovskite absorber. Type-II band alignment is generally regarded as the energetically favorable configuration for promoting efficient charge separation and carrier extraction in MoS₂/perovskite heterojunctions [3,33,39,40]. In contrast, the type-I alignment discussed here represents a specific interfacial scenario observed in the wafer-scale monolayer MoS₂/FAPbI₃/MoS₂ architecture reported by Zai et al. [1], rather than a broadly applicable characteristic of MoS₂-modified perovskite solar cells. In this device configuration, continuous monolayer MoS₂ films are strategically introduced at both the PTAA/FAPbI₃ and FAPbI₃/C₆₀ interfaces. Ultraviolet photoelectron spectroscopy and optical absorption measurements reveal that the valence-band maxima of FAPbI₃ and monolayer MoS₂ are located at −5.37 and −5.58 eV, respectively, while their conduction-band minima are positioned at −3.85 and −3.70 eV. This energetic arrangement places the band edges of FAPbI₃ entirely within the wider bandgap of monolayer MoS₂, thereby establishing a type-I band alignment (Fig. 3a) [1].

Despite the conventional view that type-I alignment may hinder carrier extraction, the atomically thin nature of the MoS₂ interlayers fundamentally alters the interfacial transport mechanism. At both the hole- and electron-selective contacts, monolayer MoS₂ enables majority carriers to traverse the interface via quantum tunneling toward PTAA and C₆₀, respectively. Simultaneously, the band offsets act as energetic barriers that block minority carriers from reaching the opposite transport layers. This dual functionality suppresses interfacial non-radiative recombination and mitigates the associated V_{oc} loss. Because the

tunneling distance is confined to a single atomic layer, majority-carrier extraction remains efficient, consistent with transient-current measurements showing no discernible retardation of carrier transport. The effectiveness of this type-I interfacial configuration therefore depends on continuous monolayer coverage, precise atomic-scale thickness control, and defect-minimized transfer. Increased thickness, structural discontinuities, or contamination may introduce additional interfacial resistance, disrupt tunneling pathways, and ultimately compromise Short-Circuit Current Density (J_{sc}) and FF [1,41,42].

At the MoS_2 /perovskite heterojunction, charge redistribution driven by Fermi-level equilibration can further generate a space-charge region and, in turn, a built-in electric field (BIEF), which provides an additional driving force for the directional separation and drift of photogenerated carriers and thereby reduces interfacial non-radiative recombination [42,43]. Fig. 3b shows the microstructure of perovskite films heteroepitaxially grown on an MoS_2 template. Koo et al. [3] reported that the use of a mesoporous MoS_2 ETL increased electron mobility to $2.84 \times 10^{-3} \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$; the corresponding J–V curves and cross-sectional TEM images are shown in Fig. 3c. Time-resolved photoluminescence (TRPL) results (Fig. 3d) further indicate that MoS_2 induces stronger fluorescence quenching than conventional TiO_2 , consistent with more efficient interfacial carrier extraction. Overall, MoS_2 accelerates interfacial charge-transfer kinetics primarily by improving energy-level alignment and interfacial contact. At the same time, its relatively low intrinsic trap-state density and chemically stable surface can help limit recombination losses during transport [44]. Nevertheless, most of these observations were obtained under specific device structures and processing conditions and cannot be generalized uncritically to all systems.

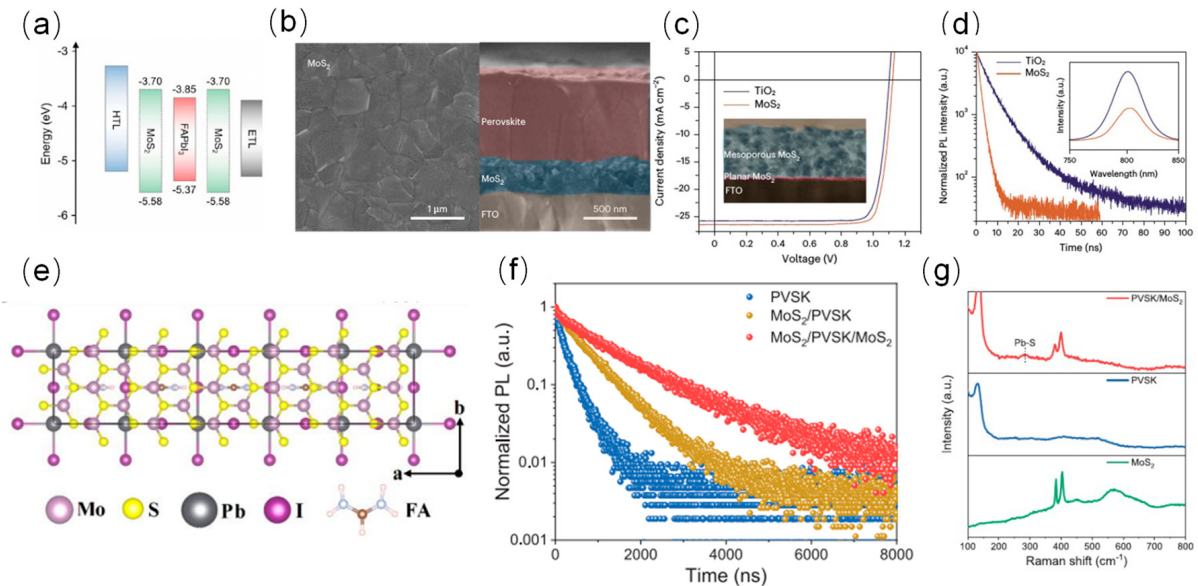


Figure 3: Interfacial energetics, morphology, carrier dynamics, and chemical interactions in MoS_2 /perovskite heterostructures. (a) Energetic band alignment at the perovskite/monolayer MoS_2 heterojunction [1]; (b) Top-view and cross-sectional SEM micrographs of perovskite thin films deposited on MoS_2 templates; (c) Photovoltaic J–V curves of the champion PSC under simulated AM 1.5G irradiation, inset: cross-sectional TEM micrograph of the FTO/planar MoS_2 /mesoporous MoS_2 architecture [3]; (d) TRPL spectra of perovskite films quenched by TiO_2 and MoS_2 substrates [3]; (e) Top-view schematic representation of the MoS_2 /FAPbI₃/ MoS_2 heterostructure; (f) TRPL decay profiles for pristine perovskite, MoS_2 /perovskite, and MoS_2 /perovskite/ MoS_2 multilayered films [1]; (g) Raman vibrational spectra of MoS_2 , perovskite, and the resultant MoS_2 /perovskite composite [1].

2.2 Synergetic Chemical Defect Passivation and Recombination Mitigation

Perovskite films commonly contain dangling bonds and ionic vacancies at their surfaces and grain boundaries, where undercoordinated Pb^{2+} species are widely regarded as the dominant trap centers responsible for non-radiative recombination. Sulfur sites on the MoS_2 surface possess lone-pair electrons and can therefore act as Lewis bases, forming coordination bonds (Pb-S) with Lewis-acidic Pb^{2+} centers. This interfacial coordination passivates dangling bonds and neutralizes deep-level traps, thereby reducing the carrier-trap density at the source [1]. Fig. 3e,f show, respectively, the top-view atomic model of the $\text{MoS}_2/\text{FAPbI}_3/\text{MoS}_2$ heterostructure and the corresponding TRPL decay curves. Experimental results indicate that introducing MoS_2 significantly prolongs carrier lifetime relative to pristine perovskite films, with the longest PL decay observed under bilateral interfacial modification. This behavior points to more effective suppression of interfacial non-radiative recombination and suggests more complete passivation of defects at the perovskite surface and interfaces. The relevant interfacial chemical interactions are further supported by X-ray photoelectron spectroscopy (XPS) and Raman spectroscopy (Fig. 3g). In addition, MoS_2 helps stabilize the black α - FAPbI_3 phase, as the interfacial coordination suppresses its transition to the non-perovskite δ Phase. Density functional theory (DFT) calculations show that the corresponding phase-transition barrier can increase by more than 50%, supporting the view that such interfacial interactions reduce defect-state density and improve phase stability [1,45]. Overall, this passivation route suppresses Shockley–Read–Hall (SRH) recombination at the interface, thereby improving both carrier lifetime and V_{oc} .

As a layered material that combines charge-transport and interfacial-regulation functions, MoS_2 can further reduce recombination losses through multiple pathways. From the standpoint of band structure, when used as an ETL or HTL, suitable valence- or conduction-band positions can help establish selective charge-blocking behavior, thereby reducing back transfer of carriers and interfacial non-radiative recombination. Beyond kinetic optimization, MoS_2 also suppresses ion migration. Time-of-flight secondary ion mass spectrometry (ToF-SIMS) depth profiling indicates that, in specific encapsulation schemes, an MoS_2 diffusion barrier can reduce the I^- signal intensity in Ag electrodes by more than 90%; Importantly, the more than 90% decrease in the I^- secondary-ion signal detected in the Ag electrode indicates markedly reduced iodine penetration under the specific test conditions, rather than a direct measurement of the ion-migration activation energy. ToF-SIMS depth profiles and Arrhenius-derived E_a values probe different aspects of ion transport and should therefore not be interpreted as quantitatively interchangeable [1]. H-type diffusion-cell experiments (Fig. 4a,b) likewise show that the MoS_2 layer can effectively impede the physical diffusion of I^- . A representative example is the work of Koo et al. [3], who further reported that a mesoporous MoS_2 network can, to some extent, hinder the ingress of moisture and oxygen, thereby slowing the hydrolysis of perovskite materials. Monolayer MoS_2 is particularly notable in this regard because its atomic-scale thickness enables interfacial passivation with little additional series resistance, helping to ease the trade-off between passivation strength and transport efficiency. In addition, owing to quantum-confinement effects, the bandgap of MoS_2 quantum dots (QDs) can widen to about 2.6 eV, allowing them to serve as interfacial buffer layers that reduce shunt losses through barrier modulation [46]. These effects, however, are often sensitive to particle size, Phase, and placement; if not properly designed, they may instead introduce additional transport barriers.

2.3 Enhancement of Perovskite Crystallinity

The formation of high-quality perovskite films is a prerequisite for high-performance PSCs. Existing studies show that the introduction of MoS_2 can markedly influence the heterogeneous nucleation kinetics of perovskites, thereby improving grain morphology and long-range order [47,48]. Specifically, MoS_2 has an

atomically smooth surface and a relatively ordered lattice framework, and it exhibits a degree of geometric compatibility with several perovskite systems. This favors van der Waals quasi-epitaxial growth at the heterointerface [3,31,47,49]. Such a process can reduce interfacial lattice strain, promote grain growth, and lower the density of grain boundaries, ultimately yielding denser films with fewer defects. Under these conditions, improvements in PCE and operational stability often accompany enhanced crystallinity rather than arising independently of it.

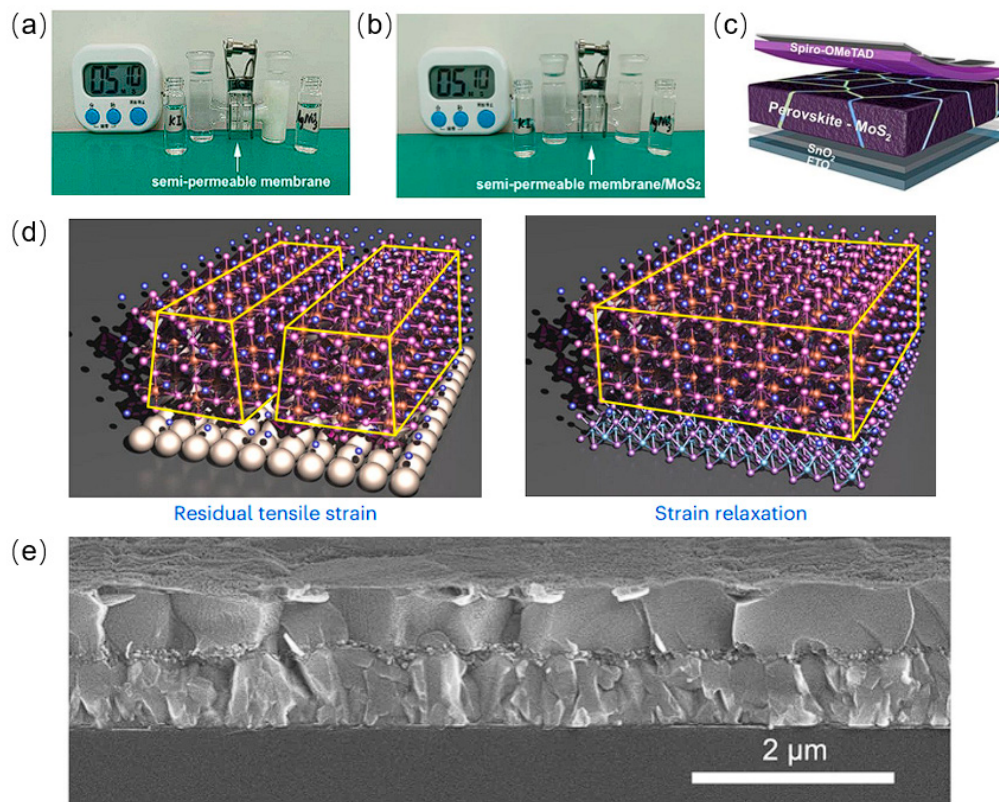


Figure 4: Experimental evidence and structural mechanisms associated with ion-diffusion blocking, crystallization regulation, strain relaxation, and interfacial stabilization by MoS₂. (a,b) Diffusion experiments conducted in an H-type diaphragm electrolytic cell [1], employing a semipermeable membrane: (a) pristine and (b) integrated with MoS₂. (c) Schematic architecture of PSCs modified with MoS₂ nanosheets [47]. (d) Schematic illustration of the residual strain distribution within perovskite films grown on TiO₂ and MoS₂ substrates [3]. (e) Cross-sectional SEM image of the device with the configuration: FTO/cTiO₂/mTiO₂/MAPbI₃/MoS₂ QDs:f-RGO/spiro-OMeTAD/Au [40].

The smooth and nearly dangling-bond-free surface of MoS₂ can serve as a van der Waals epitaxial template, promoting ordered liquid-phase epitaxy of MAPbI₃ [41,42]. Owing to lattice-symmetry compatibility, the resulting crystals tend to adopt a preferred out-of-plane (110) orientation, which is generally favorable for vertical carrier transport. A direct consequence of van der Waals epitaxy is higher perovskite crystallinity, together with a concurrent reduction in bulk and interfacial defects. When such films are incorporated into photovoltaic devices, both V_{oc} and fill factor (FF) often improve; PSCs in which MoS₂-regulated crystallization has already achieved PCE values above 25% [1,3], indicating that crystallinity control is indeed one of the key mechanisms behind performance enhancement. A representative example is the work of Zhang et al. [47] (Fig. 4c), who showed that size-tunable MoS₂ nanosheets can simultaneously modulate heterogeneous

nucleation, grain growth, and residual-strain relaxation, thereby improving the structural integrity and stability of perovskite films prepared by sequential deposition and ultimately enhancing device performance.

When MoS₂ is used as an ETL, its distinctive interfacial topology can not only induce preferred crystal orientation but also facilitate the release of residual lattice strain [3,50,51] (Fig. 4d), which is particularly important for reconciling high efficiency with long-term operational stability. Meanwhile, MoS₂ nanosheets or QDs dispersed in the precursor solution can act as heterogeneous nucleation templates, lowering the free-energy barrier for nucleation, making the distribution of nuclei more uniform, and guiding anisotropic grain growth. This process ultimately leads to denser, more continuous, and less pinhole-prone films, thereby improving FF and device reproducibility [47]. The loading concentration of MoS₂, however, often exerts a dual effect on the optoelectronic response: moderate incorporation can improve carrier extraction, whereas excessive incorporation may introduce parasitic recombination centers or cause an interfacial energy-level mismatch, thereby compromising PCE [3]. For example, Dai et al. [52] reported that increasing the concentration of an MoS₂/PAS nanocomposite in PEDOT:PSS to 3 mg mL⁻¹ could increase PCE from 14.69% to 16.47%, whereas further increasing the concentration to 5 mg mL⁻¹ led instead to performance deterioration.

2.4 Mechanisms for Stability Enhancement

The strategic integration of MoS₂ constitutes a primary strategy for fortifying the long-term operational stability of PSCs. These stabilization mechanisms are multifaceted, synergistically encompassing interfacial chemical passivation, robust physical barrier effects, and structural reinforcement.

In terms of chemical stability, the sulfur-rich surface of MoS₂ can form stable Pb–S coordination bonds with undercoordinated Pb²⁺ ions at the perovskite interface, which thermodynamically favors the stabilization of the photoactive black Phase of FAPbI₃ and suppresses conversion to non-photoactive phases as well as related phase segregation [1,3,33,53,54]. With respect to halide-ion migration, particularly I⁻ migration, MoS₂ can suppress ionic redistribution through complementary physical and chemical effects. A continuous monolayer primarily acts as an atomically thin diffusion barrier, whereas Pb–S coordination passivates interfacial defects, stabilizes the perovskite structure, and may limit defect-assisted migration pathways. However, in the absence of paired Arrhenius-derived E_a values for otherwise identical control and MoS₂-modified samples, the ToF-SIMS and diffusion-cell results should be interpreted as evidence of reduced iodine transport across the interface, rather than as quantitative proof that MoS₂ increases E_a [1,47,55,56]. As a physical barrier, the chemical inertness and relatively dense layered structure of two-dimensional MoS₂ can also impede the ingress of external moisture and oxygen to some extent, thereby improving resistance to hydrolysis and oxidation [1]. In addition, MoS₂ can strengthen interfacial structural integrity by improving energy-level alignment, relieving residual strain, and reducing interfacial defect density [1,3,57]. Together, these effects help the device retain structural stability under photostrictive, electric-field, and thermal stresses. A representative example is the MoS₂ QDs:f-RGO hybrid interfacial-modification strategy reported by Najafi et al. [40]. Cross-sectional SEM images (Fig. 4e) show that this hybrid structure better preserves film morphology during operation. After 1032 h of exposure to ambient conditions without encapsulation, devices containing MoS₂ still retained 91.2% of their initial PCE, significantly higher than the 76.4% retained by the control MAPbI₃ devices and the 88.8% retained by the f-RGO-modified counterparts. Overall, MoS₂ shows substantial promise for simultaneously improving photovoltaic efficiency and long-term operational stability in PSCs. However, the magnitude of this stabilization effect still depends on the specific interfacial design and encapsulation conditions, and it should not be assumed to act equally effectively against all degradation pathways.

3 Integration Strategies and Fabrication Methodologies of MoS₂

Section 2 discussed the main mechanisms through which MoS₂ functions in PSCs. However, whether these effects can be realized in a stable and reproducible manner depends largely on the preparation quality of the material itself and the way it is incorporated into the device. This section, therefore, turns to the fabrication and integration of MoS₂, focusing on the characteristics of different synthesis routes and on how they influence subsequent interfacial construction, process compatibility, and device performance.

3.1 Synthesis and Fabrication Techniques for MoS₂

The synthesis of MoS₂ can generally be classified into two categories based on the growth mechanism: top-down and bottom-up. The former mainly relies on external energy to weaken the weak vdW interactions between layers in bulk MoS₂ crystals, thereby exfoliating bulk precursors into mono- or few-layer nanostructures [58–60]. These methods are typically based on physical or mechanical action. Mechanical exfoliation, for example, relies primarily on adhesion-assisted cleavage and interlayer delamination to separate mono- or few-layer flakes from bulk MoS₂ crystals [61]. Building on this, chemically assisted strategies can further improve exfoliation efficiency. A representative example is ion-intercalation-assisted exfoliation [62], in which chemical intercalants first expand the interlayer spacing and weaken interlayer coupling, followed by physical stimulation, such as ultrasonication, to enable the scalable production of MoS₂ nanosheets [63]. The main advantage of this class of methods lies in its relatively direct processing route and its ability to yield thin-layer materials with high crystallinity. At the same time, however, layer-number uniformity, production yield, and reproducible downstream processing often remain limited, and their applicability to large-area devices should therefore not be overstated. By contrast, bottom-up approaches follow an atom- or molecule-by-molecule assembly route, using precursor species as the basic building blocks to achieve layer-by-layer growth of MoS₂ nanostructures or continuous films through controlled chemical reactions or physical deposition [64]. Common methods include chemical vapor deposition (CVD) [65], hydrothermal/solvothermal synthesis [66,67], and atomic layer deposition (ALD) [68,69], all of which generally require relatively strict control over reaction kinetics. In addition to these chemical routes, physical vapor deposition (PVD) represents another bottom-up approach for preparing continuous ultrathin MoS₂ films at relatively low temperatures [70]. Overall, bottom-up strategies are more favorable for obtaining functional layers with good continuity and are more promising for device-level integration. Even so, clear trade-offs remain among crystallinity, product size, phase control, and process temperature. A detailed comparison of the preparation conditions, advantages, and limitations of these methods is provided in Table 1.

Table 1: Comparison of representative MoS₂ fabrication methods relevant to integration in perovskite solar cells.

Preparation Method	Typical Conditions	Advantages	Disadvantages
Top-down—Mechanical Exfoliation [61]	Adhesive-assisted cleavage of bulk MoS ₂ crystals, followed by optical or spectroscopic identification of mono- and few-layer flakes.	Produces highly crystalline flakes with low defect densities; suitable for fundamental studies and device prototyping	Low yield; difficult to scale for large-area continuous films
Top-down—Ion Intercalation Exfoliation [62]	Chemical or electrochemical ion intercalation, followed by solvent-assisted delamination or ultrasonication.	Higher yield than mechanical exfoliation; produces solution-processable mono- or few-layer dispersions	May introduce residual intercalants, structural defects, and 2H-to-1T/1T' phase conversion; layer-number and phase uniformity remain difficult to control

Table 1: *Cont.*

Preparation Method	Typical Conditions	Advantages	Disadvantages
Bottom-up—CVD [65] (Chemical Vapor Deposition)	Vapor-phase reaction of Mo- and S-containing precursors at elevated temperatures, followed by nucleation and lateral growth on a selected substrate.	Large-area continuous-film growth, relatively good crystallinity, and controllable layer number under optimized growth conditions.	Elevated temperature, substrate dependence, grain boundaries, nucleation-density control, and possible transfer-induced contamination or damage.
Bottom-up—Hydrothermal [66]/ Solvothermal [67]	Mo-containing and S-containing precursors are reacted in a sealed autoclave at elevated temperature to form MoS ₂ nanosheets, particles, or hierarchical structures.	Low-cost batch synthesis with relatively high yield and compatibility with solution processing	Limited control over phase composition, lateral size, thickness distribution, stoichiometry, and morphology; does not directly produce continuous device-grade films
Bottom-up—ALD [68,69] (Atomic Layer Deposition)	Self-limiting cyclic reactions using volatile Mo and S precursors at low-to-moderate temperatures; post-deposition annealing or sulfurization is often required to improve crystallinity.	Excellent conformality, thickness uniformity, and wafer-scale compatibility	Slow deposition rate; expensive equipment; difficult to achieve high crystallinity
Bottom-up—PVD [70] (Physical Vapor Deposition)	Sputtering or evaporation of Mo or MoS ₂ source material onto a substrate	Mature vacuum-processing technology with controllable film thickness, relatively high deposition rates, and good large-area coverage	As-deposited films may exhibit low crystallinity, sulfur deficiency, non-stoichiometry, and high defect densities; post-annealing or sulfurization is often required.
Bottom-up—Molecular Beam Epitaxy (MBE) [71]	Epitaxial growth using controlled atomic or molecular beams under ultrahigh vacuum and elevated substrate temperature	High purity; precise control of thickness, composition, and interface quality; suitable for high-quality monolayer films	Expensive equipment; slow growth rate; stringent vacuum and substrate requirements; limited large-area scalability
Bottom-up—Pulsed Laser Deposition (PLD) [72]	Pulsed-laser ablation of a MoS ₂ target under vacuum or a controlled atmosphere, followed by deposition on a heated substrate	Relatively rapid deposition; flexible process control; good compositional transfer from the target; suitable for crystalline thin films	Limited coating area; possible droplets and particulates; thickness nonuniformity; expensive equipment and limited industrial scalability

3.2 Integration Strategies and Fabrication Processes of MoS₂

In PSCs, MoS₂ can be incorporated in multiple ways, and different integration routes correspond to distinct functional locations and performance priorities. Based on the current literature, the most representative strategies include large-area transfer integration of monolayer films, the use of MoS₂ in interfacial engineering and passivation layers, and bulk-additive incorporation. None of these strategies is universally superior; what matters more is whether a given route matches the target device architecture, fabrication workflow, and functional requirements.

- (1) Wafer-scale integration of monolayer MoS₂ films: In recent years, the large-area integration of monolayer MoS₂ with perovskite absorbers has been regarded as one of the more important advances

in this field. In common p–i–n architectures, high-crystallinity monolayer MoS₂ can be introduced at the perovskite interface through precision transfer processes, thereby forming heterostructures such as “MoS₂/perovskite” or “MoS₂/perovskite/MoS₂” [1,73]. Fig. 5a–c illustrates the synthesis of MoS₂ and the subsequent wafer-scale transfer procedures. To date, several coordinated integration schemes have been proposed. For example, a transfer strategy assisted by thermal release tape (TRT) enables high-fidelity transfer of monolayer MoS₂ onto the PTAA/perovskite interface while largely avoiding the mechanical damage and chemical contamination commonly associated with conventional wet-transfer processes [1]. In addition, uniform deposition of MoS₂ nanosheets by Electro spray can not only improve interfacial charge transport but also enhance interlayer mechanical interlocking, thereby increasing the structural reliability of flexible devices under complex deformation conditions [46]. Although this route offers clear advantages in interfacial quality and structural integrity, its process window, transfer cost, and large-area uniformity remain practical issues that cannot be ignored.

- (2) Interfacial engineering and passivation layer applications: Another common route is to use MoS₂ directly in interfacial engineering, for example, by constructing mesoporous MoS₂ ETLs or by introducing it as a passivation or buffer layer between the perovskite and the transport layer [3,55]. Fig. 5d shows a schematic of the process for preparing mesoporous MoS₂ by a hard-template method. Compared with conventional ETLs such as TiO₂, MoS₂ typically does not exhibit pronounced photocatalytic activity under ultraviolet irradiation and can therefore, to some extent, avoid the accelerated decomposition of organic cations. At the same time, MoS₂ is often introduced as an interfacial passivation layer between the perovskite absorber and the charge-transport layers [33,73,74]. Its atomically smooth surface and near-absence of dangling bonds facilitate the construction of high-quality vdW heterointerfaces with the perovskite layer. The core function of this strategy lies in more effective passivation of defects—particularly charged ionic defects—at the perovskite surface and within GBs. As interfacial charge-trapping centers are reduced, non-radiative recombination losses decrease accordingly, and both carrier diffusion length and lifetime typically improve [20]. It should be noted, however, that these advantages of interfacial engineering often depend on the continuity of thin-layer coverage, the controllability of thickness, and compatibility with adjacent layers; if process control is insufficient, the interfacial layer itself may become a new transport barrier.
- (3) Bulk-doping integration strategy: In addition to interfacial incorporation, directly adding MoS₂ QDs or nanosheets into the perovskite precursor solution as additives during film formation is also a relatively simple and widely used integration route [47,75]. During dynamic crystallization, these nanomaterials tend to distribute preferentially in GB regions, where they reduce trap-state density by passivating dangling bonds, thereby improving the micromechanical stability of the film to some extent [45,48,73,76]. For example, Liu et al. [77] introduced few-layer MoS₂ into the CH₃NH₃PbI₃ absorber to form an in situ bulk heterostructure, raising the PCE from 15.29% to 18.31%; the resulting devices also retained about 87% of their initial efficiency after 480 h of aging. Overall, when used as a bulk additive, MoS₂ can often contribute simultaneously to crystallization control and improved charge transport. This route, however, is sensitive to loading amount, dispersion state, and compatibility with the precursor chemistry; excessive incorporation or poor dispersion may instead introduce new recombination centers or disrupt film uniformity.

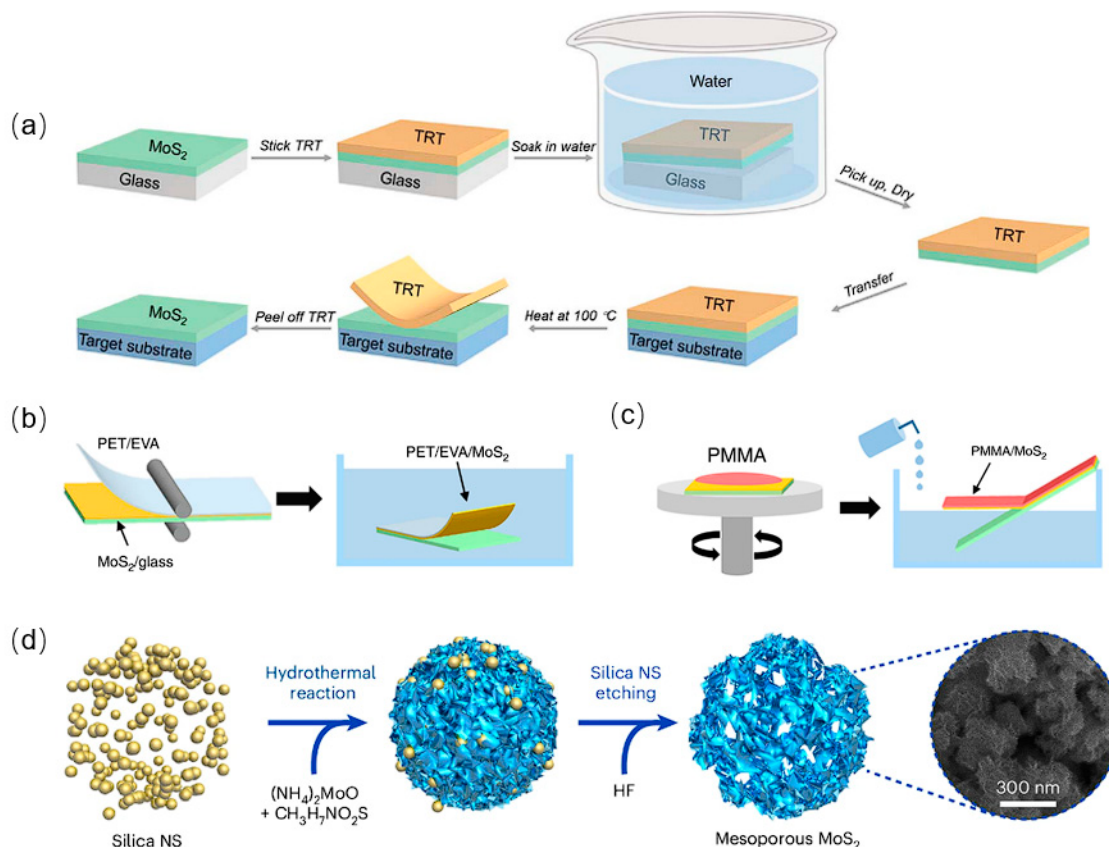


Figure 5: Schematic illustration of representative fabrication and integration methodologies for MoS₂: (a) Schematic representation of the MoS₂ thin-film transfer protocol [1]. (b) Roll-to-roll (R2R) transfer strategy for the continuous integration of MoS₂ films onto flexible substrates [73]. (c) PMMA-assisted etching-free transfer of MoS₂ onto rigid substrates, facilitating high-fidelity interfacial integration [73]. (d) Schematic depiction of the hard-template-mediated synthesis route for mesoporous MoS₂ architectures [3].

4 Applications of MoS₂ in PSC

The role of MoS₂ in PSCs is not determined solely by the material, but primarily by the functional layer in which it is incorporated and the corresponding interfacial configuration. In current studies, MoS₂ has been integrated as an ETL, HTL, interfacial layer, or additive in the absorber, where it exhibits distinct regulatory effects depending on its location. Fig. 6 summarizes representative integration configurations for MoS₂. The following sections discuss its applications across different functional layers based on these typical configurations, along with the boundaries of their applicability.

4.1 Applications as Electron Transport Layers

In conventional n-i-p architectures, MoS₂ has emerged as a representative low-temperature ETL candidate that can, under appropriate conditions, replace TiO₂. Its advantages arise mainly from favorable energy-level alignment, high specific surface area, and a relatively chemically inert surface. In mesoporous configurations in particular, MoS₂ can not only enlarge the effective heterojunction contact area but also relieve interfacial lattice mismatch and residual strain, thereby inducing preferred orientation of perovskite crystals and simultaneously improving charge separation and collection efficiencies [3,40,78].

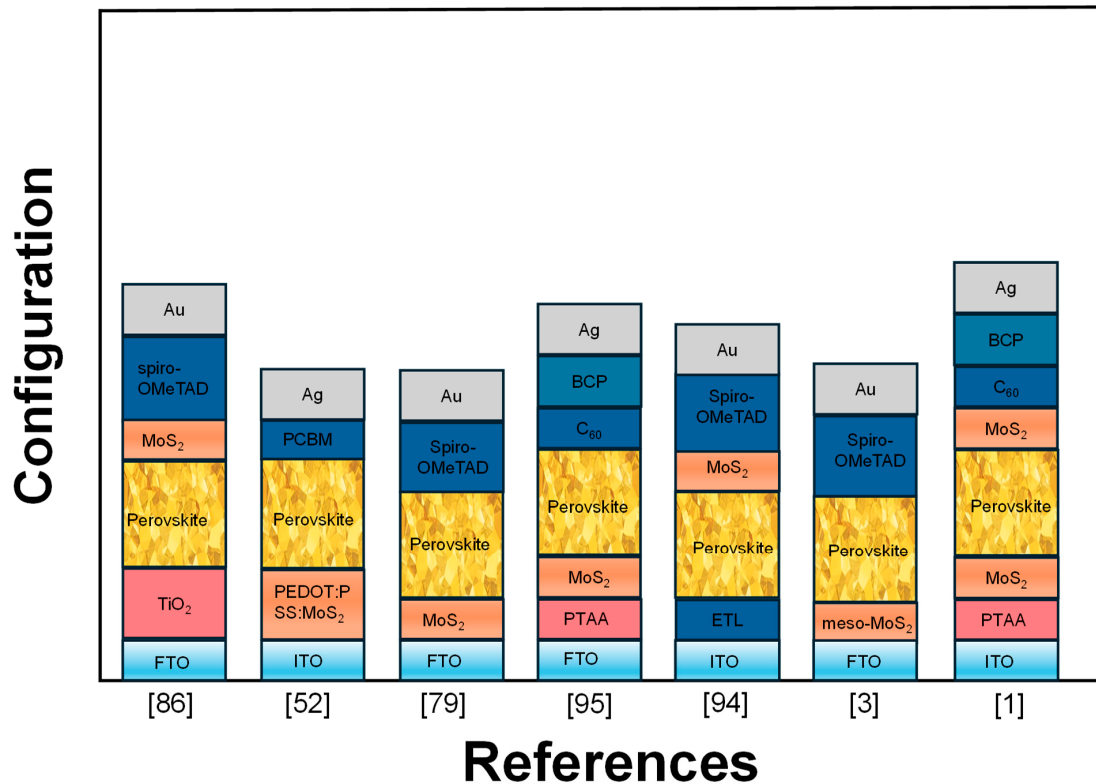


Figure 6: Schematic diagram of typical MoS₂-based configurations used to enhance the performance of perovskite solar devices. Various MoS₂ incorporation strategies, including charge-transport layers, interfacial buffer layers, and dopants within the perovskite bulk, are illustrated across different device architectures. The references for each configuration are also provided.

This rationale has been validated through different processing routes. Mahmood et al. [46] fabricated an MoS₂ nanosheet ETL by electrospray deposition and achieved a PCE above 16.17%, showing that MoS₂ can facilitate electron extraction while also serving as a template for the growth of high-quality perovskite films. To address the limitations of conventional oxide ETLs in low-temperature process compatibility and interfacial extraction efficiency, Singh et al. [79] grew transparent MoS₂ films directly on glass/FTO substrates via microwave irradiation, achieving a PCE of 13.1%. More recently, Koo et al. [3] established a mesoporous MoS₂ ETL that effectively suppressed interfacial non-radiative recombination, delivering PCEs of 25.7% and 22.4% for device areas of 0.08 cm² and 1.00 cm², respectively, clearly outperforming TiO₂-based controls. However, MoS₂ performance as an ETL remains highly sensitive to fabrication parameters. Poor control of substrate temperature may lead to non-uniform particle distribution and increased surface roughness, thereby introducing substantial leakage losses. Although optimized MoS₂ films can achieve an electron mobility of about 70 cm² V⁻¹ s⁻¹, further suppression of deep-level defects and interfacial recombination centers remains a key bottleneck for continued PCE improvement [34,80]. The feasibility of using MoS₂ as an ETL has therefore been demonstrated, but whether its advantages can be sustained at larger areas and within a more robust processing window remains to be verified.

4.2 Applications as Hole Transport Layers

In p-i-n architectures, the limited chemical stability of conventional organic HTLs, such as PEDOT:PSS, has motivated the exploration of MoS₂ as either an inorganic alternative or an interfacial regulatory

component. Existing studies show that by tuning sulfur vacancies, the work function (Φ) of MoS₂ can be increased from 4.7 eV to 5.3 eV, thereby improving alignment with the valence-band maximum (VBM) of the perovskite and enabling more effective hole extraction [60,81]. It is also worth noting that, before MoS₂ was systematically explored in PSC hole-extraction design, related studies in organic solar cells had already demonstrated the value of molybdenum sulfide- and sulfur-containing molybdenum oxide-based interlayers for anode-side energetics regulation and hole-selective transport—for example, Qin et al. [82,83] reported an *in situ* grown double-layer MoO₃/MoS₂ film as a hole-transport layer in organic solar cells, while sulfur-doped molybdenum oxide was also shown to function effectively as an anode interfacial layer. Although these studies were not conducted in PSCs, they provided useful early precedents for the later development of MoS₂-related HTL and interface-regulation strategies in perovskite devices. MoS₂ alone, however, does not always reconcile conductivity, interfacial contact, and environmental stability. As a result, its integration with organic transport materials has gradually emerged as a more practical and widely adopted strategy.

For example, Wang et al. [84] incorporated MoS₂ nanoflakes into PEDOT:PSS to construct a hybrid hole transport layer (Fig. 7a). This strategy provided a faster hole-extraction pathway, reduced interfacial recombination, and lowered electrode polarization and hysteresis. As a result, device efficiency increased by about 18.5%, recombination resistance increased by 50%, and the modified devices retained more than 95% of their initial PCE after 4 weeks. Beyond hybrid hole-transport layers, hybrid interfacial strategies have also shown clear advantages in accelerating carrier dynamics. Najafi et al. [40] constructed a hybrid interfacial layer composed of MoS₂ QDs and functionalized reduced graphene oxide (f-RGO) (Fig. 7b) to regulate the heterojunction interface in MAPbI₃-based devices. By combining the high conductivity of f-RGO with the interfacial passivation effect of MoS₂ QDs, this system promoted directional hole extraction while optimizing the energy-level gradient (Fig. 7c). It suppressed interfacial exciton recombination kinetically, leading to a PCE above 20% and improved long-term operational stability. Overall, the advantages of MoS₂ in HTL-related applications lie mainly in its tunable work function, interfacial chemical stability, and synergistic regulation within composite transport layers. However, these strategies often depend on defect-state control, blending ratio, and interlayer contact quality; if the interface is not sufficiently optimized, MoS₂ may instead limit hole extraction because of inadequate lateral conductivity or local energy-level mismatch.

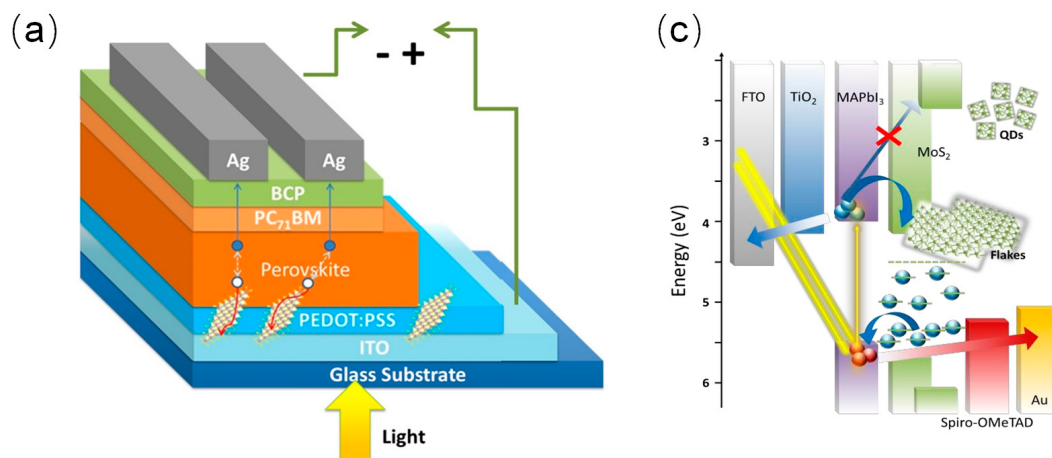


Figure 7: Cont.

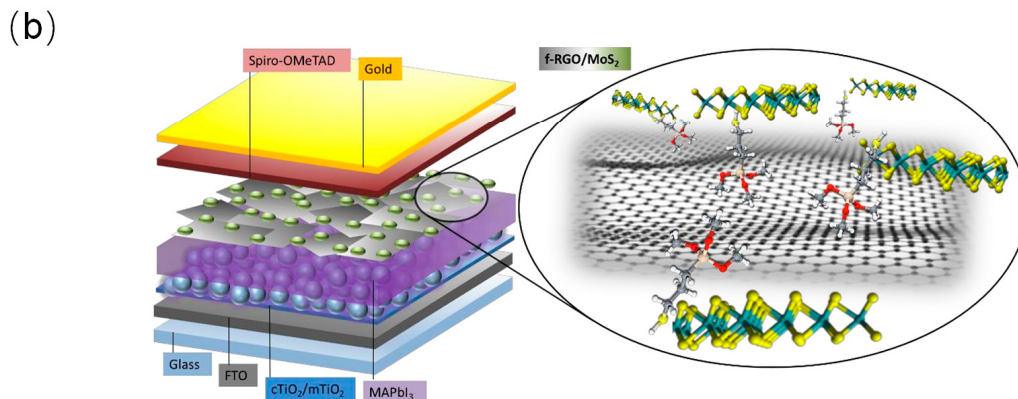


Figure 7: Representative MoS₂-based hole-transport and hybrid interfacial configurations in PSCs. **(a)** Schematic illustration of PSCs employing MoS₂-doped PEDOT: PSS as the HTL [84]. **(b)** Schematic diagram of mesoscopic MAPbI₃-based PSCs incorporating a MoS₂ QDs:f-RGO hybrid as the dual-functional HTL and anode buffer layer (ABL) [40]. **(c)** Energy band alignment diagram of the constituent materials within the assembled mesoscopic MAPbI₃-based PSCs [40,85].

4.3 Applications as Interfacial Modification Layers

Beyond its use as an independent functional layer, MoS₂ is also frequently introduced at key interfaces as a modification layer. The central aim of this route is not to replace the original transport layer, but to more precisely regulate interfacial energy levels, defect states, and ion migration while leaving the main device architecture largely unchanged. Compared with replacing an entire functional layer, interfacial modification is generally more compatible with existing processing routes and more readily integrated into established device structures. Capasso et al. reported that solution-processed few-layer MoS₂ flakes can be inserted between the perovskite and Spiro-OMeTAD as an active buffer layer [86]. This interfacial layer serves both protective and hole-transport functions, suppressing shunt contact formation between the perovskite and the Au electrode while improving interfacial charge extraction, thereby enhancing device stability.

In inverted p–i–n architectures, MoS₂ interlayers exhibit similar passivating and protective effects. On the one hand, MoS₂ can optimize vdW interactions at the perovskite/HTL interface and reduce interfacial energy loss; on the other hand, it can act as a diffusion barrier, suppressing the penetration of hygroscopic additives, such as Li-TFSI, from the HTL into the perovskite layer [87]. Zai et al. [1] reported in *Science* (2025) a dual-sided passivation strategy in which wafer-scale monolayer MoS₂ was integrated onto both the upper and lower surfaces of the perovskite by a high-fidelity transfer process. The resulting devices delivered a PCE of 26.2%, with only ~4% loss in performance after 1200 h of aging. In addition, MoS₂/graphene hybrid interfacial layers can further improve charge extraction efficiency and enhance the mechanical reliability of flexible devices under cyclic bending [33,40]. Representative examples also include the work of Xiao et al. [88], who reported simple ball-milled molybdenum sulfide nanosheets as an effective interfacial passivation material for PSCs, showing that MoS_x-type nanosheets can simultaneously reduce interfacial defects and improve device stability through a self-repairing passivation effect. This study is particularly relevant because it highlights that the value of MoS₂-related materials extends beyond charge transport to include dynamic interfacial healing and defect regulation. Overall, the value of MoS₂ as an interfacial modification layer lies in its ability to improve both electrical and chemical stability at the interface without substantially altering the main device architecture. These strategies, however, are highly sensitive to layer thickness, coverage continuity, and placement; non-uniform distribution or local aggregation of the interfacial layer may likewise introduce additional series resistance and transport barriers. Its advantages

have therefore been demonstrated quite clearly, but whether they can be reliably transferred across different device systems still needs to be assessed under specific processing conditions.

4.4 Applications as Active Layer Additives

Beyond its use as a functional layer or interfacial modifier, MoS₂ can also be introduced directly into the perovskite absorber, where it improves device performance by regulating bulk crystallization and defect states. This route does not primarily rely on additional interfacial construction; instead, it participates in film formation as an additive. Its effects are therefore manifested more directly in the coupled regulation of grain growth, GB passivation, and carrier transport.

Recent studies have extended this additive strategy from nanosheets to quantum dots. Banoth et al. [48] showed that MoS₂ nanoflakes promote perovskite grain growth and passivate defects within the grains and at grain boundaries. Yin et al. [89] further thiol-functionalized MoS₂ nanosheets into FAPbI₃, improving crystal growth and charge transport and increasing the PCE from 20.6% to 22.5%. Kambley et al. incorporated MoS₂ QDs into FAPI to form a type-I composite absorber, increasing the PCE from 12.59% to 15.01% and improving the stability of the composite film [90]. Sadhu et al. subsequently incorporated MoS₂ QDs into MAPbI₃ and observed consistent increases in photocurrent and average PCE in devices fabricated independently in two laboratories [91].

In summary, when used as an additive in the absorber, MoS₂ can simultaneously influence crystallinity, non-radiative recombination, and microstructural stability, making this route attractive for improving overall device performance. At the same time, however, it is highly sensitive to loading amount, particle-size distribution, and surface state. Moderate incorporation can improve film quality and carrier transport, whereas excessive loading, poor dispersion, or incompatibility with the precursor chemistry may instead introduce new recombination centers or disrupt film uniformity. For this reason, the next stage of development should focus less on simply increasing additive content and more on coordinated optimization of these parameters through surface functionalization, composite additive systems, and directed doping design [92–94].

4.5 Applications in Inverted Device Architectures

In Sections 4.2 and 4.3, several representative studies on inverted PSCs have already been discussed in the contexts of MoS₂-based HTLs and interfacial modification layers, and these examples will therefore not be repeated here. The p–i–n architecture is nevertheless discussed separately because many representative recent results have been reported in this system. Compared with conventional n–i–p devices, inverted PSCs generally involve lower processing temperatures, weaker J–V hysteresis, and better process compatibility with flexible devices and large-area fabrication, making them a suitable platform for interfacial engineering. This, however, does not mean that inverted architectures are intrinsically superior, as their performance remains limited by factors such as interfacial energy-level alignment, charge selectivity, and interlayer coupling. Against this background, the following section highlights representative inverted-device studies that have not been specifically discussed above.

Liu et al. [77] constructed a CH₃NH₃PbI₃:MoS₂ heterojunction and improved charge extraction at the perovskite/C₆₀ interface, increasing the device PCE from 15.29% to 18.31% while also enhancing the hydrophobicity of the perovskite layer. For flexible inverted devices, Karimipour et al. [94] combined MoS₂ nanosheets with a small-molecule HTL to form an interfacial layer that not only optimized carrier-transport kinetics but also provided stress buffering, allowing the devices to retain good photovoltaic performance under cyclic bending. To further optimize the hole-extraction interface, Dong et al. [92] introduced MoS₂ nanoparticles at the ITO/NiO_x interface and found that they could improve the intrinsic conductivity of

NiO_x by increasing the Ni³⁺: Ni²⁺ ratio while optimizing energy-level alignment. This suppressed interfacial recombination, increasing the PCE from 20.25% to 21.42%. In addition, Hu et al. [95] addressed the high interfacial resistance and slow hole extraction in PTAA by constructing a bilayer HTL composed of MoS₂ nanosheets and PTAA, thereby increasing the PCE from 14.48% to 18.47%. After storage at room temperature for 500 h, the device still retained 66% of its initial efficiency, significantly outperforming single-layer PTAA (47%) and the MoS₂-only control. Overall, the role of MoS₂ in inverted architectures is better understood as that of a tunable interfacial component rather than a simple material substitute. On the one hand, it can form more adjustable interfacial relationships with commonly used layers such as NiO_x, PTAA, and C₆₀, thereby improving selective charge transport; on the other hand, it can also enhance moisture tolerance, thermal stability, and mechanical reliability. Even so, the resulting performance gains remain closely tied to the mode of interfacial construction, material compatibility, and the matching of the overall layer stack. A more balanced view, therefore, is not to regard MoS₂ as a universal solution for inverted PSCs, but rather as an interfacial engineering material that still requires careful design.

5 Comparison of MoS₂ Integration Strategies in PSCs

Sections 2–4 discuss the mechanisms, preparation methods, and device roles of MoS₂ in PSCs. To provide a clearer comparison, the reported results are summarized in two complementary tables. Table 2 compares representative devices in terms of architecture, MoS₂ form and functional position, integration method, photovoltaic performance, and stability. Table 3 compares the thickness controllability, interfacial transport characteristics, scalability, mechanical flexibility, advantages, and limitations of different MoS₂ forms. Because stability protocols vary considerably among studies, the values in Table 2 should be interpreted together with the corresponding encapsulation, atmosphere, temperature, illumination, and operating conditions.

5.1 Coupling among Efficiency, Stability, and Scalability

Based on the available results, the most noteworthy aspect of MoS₂ in PSCs is not the local improvement of any single performance metric, but its influence on the coupling among efficiency, stability, and scalability. Early studies often treated these three aspects separately: efficiency gains were attributed mainly to improved interfacial charge extraction and energy-level alignment, stability gains were linked more often to defect passivation and diffusion blocking, whereas large-area or flexible applications were considered a later-stage extension. More recent studies, however, increasingly indicate that these three aspects are not independent. Effective suppression of interfacial defects and ion migration can not only reduce non-radiative recombination and increase PCE, but can also affect structural retention during long-term operation. Likewise, strategies that maintain film uniformity and interfacial continuity under area scaling often determine whether efficiency gains can be stably preserved. This means that the value of MoS₂ should not be reduced to an isolated contribution to either efficiency or stability, but should instead be evaluated in terms of whether it can alleviate the conventional tension among efficiency, stability, and scalability. In particular, once PSCs have entered the high-efficiency regime, further device improvement increasingly depends on the coordinated regulation of buried interfaces, surface defects, and transport bottlenecks rather than on the optimization of any single material parameter. In this sense, the role of MoS₂ has evolved from early functional validation toward a more systematic reconstruction of the device interfacial microenvironment.

Table 2: Comparison of representative MoS₂-integrated perovskite solar cells in terms of device configuration, MoS₂ form and integration role, fabrication/integration method, photovoltaic performance, and reported stability.

Year	MoS ₂ Form/Integration Role	Configuration	Fabrication/Integration Method	PCE (%)	V _{oc} (V)	FF (%)	Reported Stability	Ref.
2016	Few-layer MoS ₂ flakes; perovskite/HTL buffer layer	FTO/TiO ₂ /CH ₃ NH ₃ PbI ₃ /MoS ₂ /spiro-OMeTAD/Au	Liquid-phase exfoliation followed by solution deposition of the MoS ₂ buffer layer	13.30	0.93	66.7	Retained 93% of the initial PCE after 550 h of storage; the corresponding device without MoS ₂ retained 66%	[86]
2017	Chemically modified MoS ₂ ; blended with PEDOT as a hybrid HTL	ITO/PEDOT:PSS:modified-MoS ₂ /perovskite/PCBM/Ag	Chemical modification of MoS ₂ , blending with PEDOT, and spin coating	16.47	~1.10	75.0	Not reported quantitatively	[52]
2018	MoS ₂ QDs:f-RGO hybrid; perovskite/HTL interfacial layer	FTO/c-TiO ₂ /mp-TiO ₂ /CH ₃ NH ₃ PbI ₃ /MoS ₂ QDs:f-RGO/spiro-OMeTAD/Au	Liquid-phase exfoliation, preparation of MoS ₂ QDs, hybridization with f-RGO, and solution coating	20.12	1.11	79.75	Retained 91.2% of the initial PCE after 1032 h of aging; the atmosphere and encapsulation conditions were not specified in the reported comparison	[40]
2019	MoS ₂ film; ETL	FTO/MoS ₂ /CH ₃ NH ₃ PbI ₃ /po-Spiro-OMeTAD/Au	One-step microwave-assisted direct growth of MoS ₂ on FTO	13.1	0.89	63.0	No device-aging test reported	[79]
2020	MoS ₂ nanosheets; ETL	FTO/MoS ₂ /CH ₃ NH ₃ PbI ₃ /spiro-OMeTAD/Au	Preparation of MoS ₂ nanosheets followed by electrospray deposition	16.17	NR	NR	Not reported quantitatively	[46]
2020	MoS ₂ nanoparticles; buried buffer layer and absorber additive	ITO/PEDOT:PSS/MoS ₂ /CH ₃ NH ₃ PbI ₃ :MoS ₂ /PCBM/Bphen/Ag	Solution deposition of the MoS ₂ buffer layer and incorporation of MoS ₂ into the perovskite precursor	18.31	NR	NR	Retained 87% of the initial PCE after 20 d under continuous one-sun illumination in air	[77]
2022	Size-controlled MoS ₂ nanosheets; crystallization-regulating additive	FTO/c-TiO ₂ /mp-TiO ₂ /perovskite/spiro-OMeTAD/Au	Incorporation of size-controlled MoS ₂ nanosheets into the PbI ₂ precursor during sequential deposition	22.50	NR	NR	Retained 87% after 1200 h of ambient storage, 89% after 600 h of thermal aging, and 85.1% after 73 h of light soaking	[47]
2022	MoS ₂ nanosheets; PTAA/MoS ₂ double HTL	FTO/PTAA/MoS ₂ /CH ₃ NH ₃ PbI ₃ /C ₆₀ /BCP/Ag	Liquid-phase exfoliation and sequential spin coating of PTAA and MoS ₂	18.47	1.10	71.9	Retained 66% of the initial PCE after 500 h of storage at room temperature; the PTAA-only device retained 47%	[95]

Table 2: *Cont.*

Year	MoS ₂ Form/Integration Role	Configuration	Fabrication/Integration Method	PCE (%)	V _{oc} (V)	FF (%)	Reported Stability	Ref.
2022	Thin MoS ₂ nanosheets; perovskite/HTL interface modifier	PET/ITO/ETL/Cs _{0.05} (FA _{0.83} MA _{0.17}) _{0.95} PbI ₃ /MoS ₂ /Spiro-OMeTAD/Au	Perovskite surface treatment with 1-dodecanethiol followed by drop casting of dispersed MoS ₂ nanosheets	17.70	1.04	74.0	Recovered 95% of the maximum PCE during unencapsulated light-cycling tests and fully recovered the initial photovoltaic parameters after 300 bending cycles	[94]
2023	Few-layer MoS ₂ ; MoS ₂ /spiro-OMeTAD hybrid HTL	FTO/c-TiO ₂ /mp-TiO ₂ /CH ₃ NH ₃ PbI ₃ /MoS ₂ /spiro-OMeTAD/Ag	Liquid-phase exfoliation followed by solution deposition of the MoS ₂ -containing HTL	9.50	NR	NR	A relative PCE degradation of 45% was reported after 120 h; unencapsulated, RH = 40–50%	[96]
2024	Vertical MoS ₂ nanoflakes; interfacial scaffold and component of the MoS ₂ /CH ₃ NH ₃ PbI ₃ hybrid active region	FTO/c-TiO ₂ /m-TiO ₂ /vertical MoS ₂ nanoflakes/CH ₃ NH ₃ PbI ₃ /FGO-based HTL/metal electrode	Thermal-CVD integration of vertical MoS ₂ nanoflakes with the CH ₃ NH ₃ PbI ₃ absorber, followed by incorporation of an FGO-based hole-transport layer	15.60	NR	NR	Retained 89% of the initial PCE after 500 h of ambient aging under 1-sun illumination	[48]
2024	MPA-functionalized MoS ₂ nanosheets; crystallization regulator	FTO/SnO ₂ /FAPbI ₃ –MoS ₂ /spiro-OMeTAD/Au	Thiol functionalization of MoS ₂ nanosheets followed by incorporation into the FAPbI ₃ precursor	22.50	NR	NR	Retained 91% of the initial PCE after more than 1600 h of unencapsulated shelf storage.	[89]
2024	MoS ₂ QDs; FAPbI ₃ composite absorber	FTO/SnO ₂ /FAPbI ₃ :MoS ₂ QDs/spiro-OMeTAD/Au	Preparation of MoS ₂ QDs, blending with the FAPbI ₃ precursor, and spin coating	15.01	0.94	65.0	No quantitative device stability data were reported; improved film stability was observed.	[90]
2024	Monolayer MoS ₂ ; growth surface for a transparent Au electrode	ITO/MeO-2PACz/perovskite/C ₆₀ /monolayer MoS ₂ /Au	Integration of a monolayer MoS ₂ surface followed by deposition of a nanometre-thick transparent Au electrode	12.50	NR	NR	Not reported	[97]
2025	Mesoporous MoS ₂ ; ETL	FTO/meso-MoS ₂ /FA _{0.95} Cs _{0.05} Pb(I _{0.85} Br _{0.15}) ₃ /Spiro-OMeTAD/Au	Synthesis of mesoporous MoS ₂ followed by solution deposition as the ETL	25.7	1.16	84.8	Stable for more than 2000 h under continuous illumination.	[3]
2025	Continuous monolayer MoS ₂ ; dual-interface buffer layers	ITO/PTAA/MoS ₂ /FAPbI ₃ /MoS ₂ /C ₆₀ /BCP/Ag	Wafer-scale monolayer growth followed by high-fidelity transfer to the bottom and top perovskite interfaces	26.2	1.20	84.3	PCE loss was below 5% after 1200 h of damp-heat aging at 85°C and 85% RH	[1]

Table 2: *Cont.*

Year	MoS ₂ Form/Integration Role	Configuration	Fabrication/Integration Method	PCE (%)	V _{oc} (V)	FF (%)	Reported Stability	Ref.
2025	MoS ₂ nanoparticles; buried ITO/NiO _x interface modifier	ITO/MoS ₂ /NiO _x /perovskite/C ₆₀ /BCP/Ag	Solution deposition of MoS ₂ nanoparticles at the ITO/NiO _x interface	21.42	NR	NR	Improved thermal, light-soaking, humidity, and ambient stability was reported; quantitative retention was not explicitly stated.	[92]
2025	MoS ₂ QDs; CH ₃ NH ₃ PbI ₃ composite absorber	ITO/c-TiO ₂ /mp-TiO ₂ /CH ₃ NH ₃ PbI ₃ :MoS ₂ QDs/spiro-OMeTAD/Au	Blending of a MoS ₂ -QD dispersion with the CH ₃ NH ₃ PbI ₃ precursor followed by spin coating	10.4	NR	NR	Not reported	[91]
2025	MoS ₂ -QD-grafted MXene; perovskite/carbon interlayer	FTO/TiO ₂ /CsPbBr ₃ /MoS ₂ -MXene/carbon	Grafting of MoS ₂ QDs onto MXene followed by deposition at the perovskite/carbon interface	10.701	1.70	NR	Significantly improved device stability was reported over 50 days under persistent light irradiation and at 80% relative humidity.	[98]

Note: Photovoltaic parameters and stability results are presented as reported in the cited studies. No unreported T₈₀ or T₉₀ values were calculated from stability curves. When the original publication did not provide a numerical device-level retention value and aging duration, the result is identified as “Not reported quantitatively”. The stability data should not be ranked directly because encapsulation, atmosphere, humidity, temperature, illumination, electrical operating mode, and aging protocols differ among studies. Abbreviations: BCP, bathocuproine; Bphen, bathophenanthroline; DT, 1-dodecanethiol; ETL, electron-transport layer; f-RGO, functionalized reduced graphene oxide; HTL, hole-transport layer; MPA, 3-mercaptopropionic acid; NR, not reported; QD, quantum dot; RH, relative humidity.

Table 3: Qualitative comparison of the structural, interfacial-transport, and processing characteristics of different MoS₂ forms used in perovskite solar cells.

MoS ₂ Form	Typical Fabrication and Integration Route	Thickness Controllability	Interfacial Transport/Contact-Resistance Evidence	Fabrication Scalability	Mechanical Flexibility	Main Advantages	Main Limitations and Suitable Applications	Representative Refs.
Continuous mono-layer films	Wafer-scale growth followed by transfer to perovskite/transport-layer or transport-layer/metal-electrode interfaces	Excellent; thickness is intrinsically defined at the atomic-layer level	Minimal additional vertical transport distance when the film is continuous and residue-free. Wrinkles, cracks, and transfer residues may increase interfacial resistance. Standardized contact-resistance values are not available	Medium to high potential; currently limited by transfer yield, contamination control, cost, and defect-free large-area coverage	High intrinsic flexibility; device-level durability depends on adhesion and transfer quality	Conformal interface coverage, low optical loss, well-defined energy-level regulation, defect passivation, and ion-diffusion suppression	Transfer complexity and sensitivity to defects remain limiting; suitable for high-efficiency planar devices requiring precise interface regulation	[1,97]

Table 3: *Cont.*

MoS ₂ Form	Typical Fabrication and Integration Route	Thickness Controllability	Interfacial Transport/Contact-Resistance Evidence	Fabrication Scalability	Mechanical Flexibility	Main Advantages	Main Limitations and Suitable Applications	Representative Refs.
Few-layer films	Liquid-phase or electrochemical exfoliation followed by spin coating, drop casting, or blending with a transport layer.	Moderate; the average layer number can be adjusted, but the distribution is generally broad.	Strongly dependent on layer number, coverage, and flake overlap. Excessive thickness may increase series resistance, whereas incomplete coverage leaves recombination pathways.	High for solution processing, although continuous large-area coverage remains challenging	Potentially high; performance depends on flake overlap, adhesion, and the supporting substrate.	Low-temperature solution processing, chemical passivation, diffusion blocking, and broad device compatibility	Aggregation, nonuniform layer number, incomplete coverage, and batch-to-batch variation require careful control; suitable for buffer layers and hybrid HTLs	[52,86,95,96]
Nanosheets/nanoflakes	Hydrothermal synthesis, liquid-phase exfoliation, chemical functionalization, electrospray deposition, or precursor blending	Moderate; lateral size and concentration can be controlled, but layer-number distributions are usually broad	Governed by nanosheet size, orientation, loading, overlap, and dispersion. Aggregation or excessive loading may impede carrier transport, while direct contact resistance is rarely quantified.	High, particularly for solution coating, spraying, electrospray deposition, and precursor incorporation	High potential; device-level bending durability has been demonstrated for an interfacial nanosheet treatment.	Crystallization regulation, heterogeneous nucleation, residual-strain reduction, defect passivation, and compatibility with scalable solution processing	Performance is sensitive to lateral size, surface chemistry, concentration, and dispersion quality; suitable as absorber additives, interlayers, and solution-processed ETLs	[46–48,94]
Mesoporous MoS₂	Controlled synthesis of mesoporous MoS ₂ followed by solution deposition as an ETL	Moderate to high for pore structure and morphology; pore-wall thickness is less precisely controlled than monolayer thickness.	Provides a large interfacial area and interconnected transport pathways. Effective resistance depends on pore filling, film continuity, and perovskite/ETL contact.	Medium to high; performance has been demonstrated in a 1.00 cm ² device, but large-area coating uniformity and pore reproducibility require further validation	Not evaluated in the cited PSC study	Efficient electron extraction, increased interfacial contact, reduced residual strain, and avoidance of high-temperature mesoporous-TiO ₂ processing	Reproducible pore structure, film uniformity, and mechanical integrity remain concerns; currently most suitable for rigid n-i-p devices and enlarged-area ETLs	[3]

Table 3: *Cont.*

MoS ₂ Form	Typical Fabrication and Integration Route	Thickness Controllability	Interfacial Transport/Contact-Resistance Evidence	Fabrication Scalability	Mechanical Flexibility	Main Advantages	Main Limitations and Suitable Applications	Representative Refs.
Quantum dots	Quantum-dot synthesis or size reduction followed by absorber blending, hybridization, or interfacial deposition	High for particle-size and quantum-confinement control; deposited-layer thickness and surface coverage are less precisely controlled	Quantum confinement enables energy-level tuning, but dispersed QDs do not form a continuous transport layer. Transport depends strongly on QD loading, dispersion, and surface ligands.	High potential for solution processing; reproducibility depends on particle-size distribution, ligand chemistry, and colloidal stability	Potentially compatible with flexible processing; device-level bending durability has not been demonstrated in the cited studies.	Grain-boundary and bulk-defect passivation, tunable energy levels, possible carrier confinement, and compatibility with precursor blending	Aggregation, insulating ligands, discontinuous transport pathways, and concentration-dependent recombination may reduce performance; suitable mainly as absorber additives or ultrathin interfacial modifiers.	[40,90,91]
MoS₂-based hybrid structures	van der Waals hybridization, surface grafting, or composite formation with graphene, MXenes, MoO ₃ , or related materials	Component- and architecture-dependent; generally moderate	Can combine MoS ₂ passivation with the conductivity or surface functionality of a second component. Additional heterointerfaces may also introduce transport resistance	Medium to high potential; composition control, dispersion, and process reproducibility are more complex than for single-component MoS ₂	Component-dependent; direct device-level bending tests are generally not reported	Combines MoS ₂ defect passivation and diffusion blocking with complementary conductivity, energy-level regulation, or chemical functionality	Increased material complexity, additional interfaces, uncertain long-term compatibility, and more demanding process control; suitable for multifunctional interfaces and specialized device architectures	[40,98,99]

Note: Thickness controllability, interfacial transport, fabrication scalability, and mechanical flexibility are compared qualitatively because the cited studies did not use standardized measurement and testing protocols. Contact resistance should not be inferred directly from device series resistance, impedance-derived charge-transfer resistance, film conductivity, or carrier mobility. Therefore, numerical contact-resistance values are not provided unless they were measured directly using comparable methods. Abbreviations: ETL, electron-transport layer; HTL, hole-transport layer; QD, quantum dot.

Nevertheless, this coupling does not arise automatically. Many high-efficiency results remain concentrated in small-area devices, and interfacial regulation that is effective at that scale may not remain equally effective at the module scale. Conversely, some strategies developed for large-area or flexible devices may improve structural consistency and mechanical reliability without necessarily preserving the highest laboratory efficiencies. Accordingly, the more meaningful criterion for assessing progress in MoS₂-based PSCs is not whether a given record value has been achieved, but whether performance advantages can be maintained simultaneously across different device scales and stress conditions.

5.2 Benefits and Trade-Offs Associated with Different Forms of MoS₂

Available studies show that the role of MoS₂ depends heavily on its specific form, and the advantages of different material configurations come with distinct costs. Monolayer or few-layer MoS₂ films are generally better suited for constructing high-quality interfaces, particularly when continuous coverage, reduced parasitic series resistance, and buried-interface or dual-side passivation effects are required. Existing results indicate that when the coverage continuity and interfacial matching of monolayer MoS₂ are sufficiently controlled, its role is not limited to reducing instantaneous recombination losses, but can simultaneously improve device efficiency and structural stability under damp-heat conditions. At the same time, however, this route usually depends on high-quality growth and high-precision transfer, involves a relatively narrow process window, and is more sensitive to large-area uniformity, transfer yield, and manufacturing cost. Monolayer or few-layer MoS₂ should therefore be regarded more as a high-precision interfacial construction tool than as a universally transferable solution across all device systems. By contrast, mesoporous or nanosheet-based MoS₂ is easier to integrate into existing device processes. It shows better charge-extraction behavior and area compatibility when used in ETLs or hybrid interfacial layers, but it is also more prone to introducing morphological roughness, non-uniform coverage, or local leakage pathways. As for MoS₂ QDs, their main advantages lie in better dispersibility, more flexible modes of incorporation, and effective passivation of GBs and local interfacial defects. Their actual effects, however, depend strongly on size distribution, surface chemistry, and compatibility with the precursor environment, and slight deviations may instead generate new recombination centers.

There is therefore no absolute “best route” among the different forms of MoS₂. A more accurate interpretation is that monolayer or few-layer films function as refined interfacial construction tools, mesoporous or nanosheet systems are more relevant to process compatibility and area extension, and QDs are better suited for localized passivation and additive-based modulation. If future studies continue to ask only which form performs best, progress in the field will likely remain limited. A more meaningful question is which form is most appropriate under a given device architecture and manufacturing condition, and what trade-offs are required to reach its performance ceiling.

5.3 Practical Relevance and Manufacturability Constraints

Compared with small-area high-efficiency devices, flexible and scalable PSCs provide a more stringent test of the practical relevance of MoS₂ strategies. The key issue is no longer whether interfacial regulation can produce local performance improvements, but whether those benefits can be retained under mechanical deformation, area scaling, and more complex process chains. In this context, the value of MoS₂ should not be judged primarily by a few high-performing devices, but rather by whether it can deliver reproducible interfacial gains under conditions more relevant to manufacturing. At the same time, flexible devices and large-area/module devices should not be evaluated within the same framework. For flexible devices, the more critical questions are whether interfacial contact can be maintained under bending, whether low-temperature

processing remains compatible, and whether MoS₂ can help relieve local mechanical stress. In other words, the flexible direction is concerned more directly with interfacial contact retention, thermal budget, and stress management. By contrast, the central issue in large-area and module devices is not adaptability to deformation, but whether film uniformity is preserved after scale-up, whether series resistance and local defects are amplified, whether coating/transfer consistency can be maintained, and whether performance remains statistically consistent after encapsulation and interconnection. Treating flexible, large-area, module, and scalable devices as simple parallel categories, therefore, risks obscuring their fundamentally different evaluation criteria and overestimating interfacial gains that are valid only in a single context.

Based on the available evidence, the practical value of MoS₂ is better understood as an interfacial-enabling effect, namely, the ability to improve interfacial continuity, mitigate local defect amplification, and, to some extent, enhance process compatibility without substantially altering the overall device architecture. Emerging studies indicate that such effects are not confined to small-area devices. For example, Agresti et al. [100] co-integrated functionalized MoS₂ with graphene and Ti₃C₂T_x MXenes into module devices, achieving active-area PCEs of 17.2% and 14.7% on 121 cm² and 210 cm² substrates, respectively. This suggests that two-dimensional heterointerfaces can play a practically meaningful role in charge homogenization over large areas. However, whether such gains can be reliably transmitted across longer process chains remains insufficiently validated. In particular, the current literature has not yet established a robust comparative framework for module-scale consistency, process tolerance, manufacturing yield, encapsulation compatibility, and statistical reproducibility. For this reason, the more defensible conclusion at present is that MoS₂ has demonstrated realistic potential for flexible, scalable PSCs, but should not yet be regarded as a generally validated manufacturing solution.

What this direction now requires, therefore, is not simply a few more champion devices, but a more comparable framework for device area, testing protocols, and failure modes, so that genuinely transferable strategies can be distinguished from those that remain at the early-stage feasibility stage. Only when the interfacial benefits of MoS₂ remain statistically consistent under area scaling, process variation, and long-term operation can its true boundaries and practical value in manufacturable PSCs be more clearly defined.

6 Conclusions and Outlook

Although PSCs have achieved rapid improvements in PCE and low-cost fabrication, their further development remains limited by interfacial energy losses, insufficient long-term stability, and challenges in maintaining performance during scale-up. In this context, MoS₂ has emerged as an effective interfacial engineering material rather than merely a replacement for individual functional layers. As discussed in this review, MoS₂ can regulate interfacial charge transport, passivate defects via Pb–S coordination, suppress ion migration, control perovskite crystallization, relieve residual stress, and enhance resistance to environmental degradation. These advantages have enabled PSCs incorporating MoS₂ to achieve PCEs up to 26.2%, along with significantly improved stability under demanding conditions. However, MoS₂ should not be considered a universal solution. Its performance depends strongly on factors such as phase, thickness, lateral size, surface chemistry, coverage continuity, integration position, and compatibility with adjacent layers. In addition, variations in device architecture, processing conditions, active area, and stability testing protocols hinder direct comparison across studies. Future research should therefore shift from empirical optimization toward controllable MoS₂-based interface engineering, with greater emphasis on reproducibility, process tolerance, and consistency under manufacturing-relevant conditions.

AI-guided materials discovery and machine-learning-assisted process optimization offer promising routes to accelerate the identification of optimal MoS₂ configurations, including phase, size, defect density,

surface functionalization, integration strategy, and processing conditions for different device architectures. These approaches may also help clarify the complex relationships among material structure, interfacial energetics, film formation, device performance, and long-term stability. However, their effectiveness relies on standardized datasets that consistently report material properties, precursor chemistry, deposition parameters, device metrics, active area, interfacial characterization, and stability testing conditions. Without such standardization, machine learning models risk capturing laboratory-specific trends rather than broadly applicable design principles.

From a manufacturing perspective, the choice of MoS₂ integration strategy should be guided by the requirements of the target device platform rather than material performance alone. For flexible and wearable PSCs, the limited thermal tolerance of polymer substrates and the need to maintain interfacial integrity under repeated deformation favor low-temperature processing routes. CVD-grown monolayer MoS₂ combined with high-quality dry transfer, lamination, or roll-to-roll techniques is attractive for achieving continuous and ultrathin interfaces, as the high-temperature growth step can be decoupled from device fabrication. However, this approach requires precise control over transfer-induced defects such as residues, wrinkles, cracks, and misalignment. Alternatively, solution-processed MoS₂ nanosheets or quantum dots deposited via spray coating, electrospray, blade coating, or slot-die coating offer cost-effective and mechanically compliant options. However, their practical implementation requires careful control of dispersion stability, flake size distribution, film uniformity, thickness, and mechanical durability under cyclic bending.

For large-area rigid devices and modules, scalable liquid-phase processing of exfoliated or functionalized MoS₂ is generally more compatible with continuous coating techniques such as blade, slot-die, and spray coating. Key parameters include ink rheology, substrate wettability, drying dynamics, thickness uniformity, interfacial coverage, reproducibility, and compatibility with module interconnection and encapsulation. While ALD and PVD can provide improved conformality and thickness control, their industrial application is limited by deposition rate, equipment cost, crystallinity control, sulfur stoichiometry, vacuum requirements, and potential post-treatment steps. Similarly, although CVD-grown monolayer MoS₂ offers high interfacial quality, challenges related to transfer yield, contamination, defect formation, and large-area alignment remain significant. Accordingly, flexible devices should be evaluated based on thermal budget, interfacial adhesion, bending radius, fatigue resistance, and performance retention. In contrast, large-area modules should be assessed in terms of coating uniformity, manufacturing yield, series resistance distribution, interconnection compatibility, and area-dependent performance consistency.

Integrating MoS₂ with other two-dimensional materials may further enhance its functionality. Heterostructures incorporating graphene, MXenes, WS₂, h-BN, or black phosphorus can provide complementary advantages in charge transport, energy-level alignment, defect passivation, ion migration suppression, environmental stability, and mechanical robustness. However, their effectiveness depends on factors such as band alignment, chemical compatibility, deposition sequence, additional resistance, structural continuity, and long-term stability. Future studies should therefore systematically compare these heterostructures with single-component MoS₂ interfaces under identical conditions, rather than attributing performance improvements solely to increased material complexity.

In perovskite/silicon tandem solar cells, ultrathin MoS₂ layers may serve as transport, passivation, diffusion-blocking, or recombination-regulating layers. Their implementation must minimize parasitic absorption, interfacial resistance, optical losses, and potential damage to the underlying subcell. For flexible and wearable photovoltaics, future designs should integrate low-temperature MoS₂ processing with robust interfacial adhesion, strain-tolerant architectures, lightweight encapsulation, and stable electrical contacts under repeated mechanical stress and environmental exposure. Continued progress in these emerging

directions, combined with platform-specific manufacturing strategies and standardized evaluation protocols, will be critical for advancing MoS₂ from a laboratory-scale interfacial modifier to a scalable and reliable component in high-performance perovskite photovoltaics.

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